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single atom with a positive or negative charge resulting from the loss of gain of one or more valence electrons and behaves as a unit, yet it still has a charge. (+ or -). Determined by the amount of electrons lost or gained.

Chapter 9: Chemical Names and Formulas Flashcards | Quizlet

To name a binary molecular compound, use the following guidelines:

1. Write the names of the elements in the order listed in the formula
2. Use prefixes appropriately to indicate the number of each kind of atom
3. End the name of the second element with the suffix -ide

Chemistry Chapter 9 Chemical Names and Formulas Flashcards ...

First, confirm that the compound is a binary molecular compound. Name the elements in the order listed in the formula. Use prefixes

to indicate the number of each kind of atom. Omit the prefix -mono when the formula contains only one atom of the first element in the name. The suffix of the name of the second element is -ide.

Study Chapter 9: Chemical Names and Formulas Flashcards ...

1. when the name of the anion (X) ends in -ide the acid name begins with the prefix hydro-. The stem of the anion has the suffix -ic and is followed by the word acid (ex. HCl= hydrochloric acid)
2. when the anion name ends in -ite the acid name is the stem of the anion with the suffix -ous followed by the word acid (ex. H₂SO₃ -- X=sulfite = sulfurous acid)

CHAPTER 9 Chemical Names And Formulas Test Review ...

Writing the Names and Formulas for Bases. A base is a compound that produces OH⁻ in water.

When naming a base, you name it like any other compound that starts with a regular or transition metal. Ex: NaOH = sodium hydroxide. When writing the formula for a base, remember to balance charges. Ex: magnesium hydroxide = Mg(OH)₂

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Chapter 9 Chemical Names and Formulas

83 SECTION 9.3 NAMING AND WRITING FORMULAS FOR MOLECULAR COMPOUNDS (pages 268-270)

This section explains the rules for naming and writing formulas for binary

molecular compounds. Naming Binary Molecular Compounds (pages 268-269)

1. Circle the letter of the type(s) of elements that form binary molecular compounds.

Name Date Class

CHEMICAL NAMES AND FORMULAS

SECTION 9.2 NAMING AND WRITING FORMULAS FOR IONIC COMPOUNDS

I. Write the formulas for these binary ionic compounds.

c. potassium iodide

2. write the formulas for the compounds formed

c. b. f. 3. Name the following binary ionic compounds.

sodium sulfide

g. CuCl

h. snc.l. a. MnO₂

c. d. SrBr₂

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ion d. mercury(II) ion 45. cyanide, CN- and hydroxide, OH- 46. a. hydroxide ion b. lead(IV) ion c. sulfate ion d. oxide ion 47. zero; A compound is electrically neutral. 48. The symbols for the cation and CHAPTER 9 Study Guide - Quia Chapter 9 Chemical Names Formulas Chapter 9 - Chemical Names and Formulas. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. bibiloni1901. Terms in this set (20) Monatomic Ion. Consists of a single atom with a positive or negative charge resulting from the loss of gain of one or more valence Chapter 9 Chemical Names Formulas Quiz Answer Key Chapter 9 "Chemical Names and Formulas" Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; Use these

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the exam. :D. Terms in this set (7) monoatomic ions. Chapter 9 Vocabulary: Chemical Names and Formulas ...Chapter 9 Chemical Names Formulas Test Answers chapter-9-chemical-names-formulas-answers-page-221 1/2 Downloaded from webdisk.shoncooklaw.com on December 4, 2020 by guest Read Online Chapter 9 Chemical Names Formulas Answers Page 221 When people should go to the ebook stores, search establishment by shop, shelf by shelf, it is essentially problematic. This is why we offer the book ...Chapter 9 Chemical Names Formulas Answers Page 221 ...Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer Section MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify the charges of

monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H PTS: 1 DIF: L1 REF: p. 254Chapter 9 Practice Test - Naming and Writing Chemical FormulasChemistry (12th Edition) answers to Chapter 9 - Chemical Names and Formulas - 9.1 Naming Ions - 9.1 Lesson Check - Page 269 4 including work step by step written by community members like you. Textbook Authors: Wilbraham, ISBN-10: 0132525763, ISBN-13: 978-0-13252-576-3, Publisher: Prentice Hall Chapter 9 Chemical Names and Formulas83 SECTION 9.3 NAMING AND WRITING FORMULAS FOR MOLECULAR COMPOUNDS (pages 268–270) This section explains the rules for naming and writing formulas for binary

molecular compounds. Naming Binary Molecular Compounds (pages 268–269)

1. Circle the letter of the type(s) of elements that form binary molecular compounds.

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1. when the name of the anion (X) ends in -ide the acid name begins with the prefix hydro-. The stem of the anion has the suffix -ic and is followed by the word acid (ex. HCl= hydrochloric acid) 2. when the anion name ends in -ite the

acid name is the stem of the anion with the suffix -ous followed by the word acid (ex. H₂SO₃ -- X=sulfite = sulfurous acid)

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Naming - Chemistry *Naming Compounds*

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CHAPTER 9 Assessment 42. a. 2- b. 1+ c. 1- d. 3+ 43. a. 2+ b. 2+ c. 3+ d. 1+ 44. a. barium ion b. iodide ion c. silver ion d. mercury(II) ion 45. cyanide, CN-

and hydroxide, OH⁻ 46. a. hydroxide ion b. lead(IV) ion c. sulfate ion d. oxide ion 47. zero; A compound is electrically neutral. 48. The symbols for the cation and

Chapter 9 Chemical Names Formulas Quiz Answer Key

SECTION 9.2 NAMING AND WRITING FORMULAS FOR IONIC COMPOUNDS I.

Write the formulas for these binary ionic compounds. c. potassium iodide 2. write the formulas for the compounds formed c. b. f. 3. Name the following binary ionic compounds. sodium sulfide g. CuCl: h. snc.l. a. MnO₂ c. d. SrBr₂ 221 Chapter 9 Chemical Names and Formulas *Study Chapter 9: Chemical Names and Formulas Flashcards ...* Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer

Section MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify the charges of monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H PTS: 1 DIF: L1 REF: p. 254

Chapter 9 Chemical Names And Formulas Workbook Answers ...

Chapter 9: Chemical Names and Formulas. STUDY. PLAY. Monatomic Ion. Consists of a single atom with a positive or negative charge resulting from the loss of gain of one or more valence electrons and behaves as a unit, yet it still has a charge. (+ or -). Determined by the amount of electrons lost or gained.

CHAPTER 9 Chemical Names And Formulas Test Review ...

Chapter 9 "Chemical Names and

Formulas" Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; Use these activities to help yourself to learn the vocabulary and major topics presented in this chapter. A B; anion: any atom or group of atoms that has a negative charge:

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First, confirm that the compound is a binary molecular compound. Name the elements in the order listed in the

formula. Use prefixes to indicate the number of each kind of atom. Omit the prefix -mono when the formula contains only one atom of the first element in the name. The suffix of the name of the second element is -ide.

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Writing the Names and Formulas for

Bases. A base is a compound that produces OH^- in water. When naming a base, you name it like any other compound that starts with a regular or transition metal. Ex: NaOH = sodium hydroxide. When writing the formula for a base, remember to balance charges. Ex: magnesium hydroxide = $\text{Mg}(\text{OH})_2$