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Directions: Questions 1–3 are long free-response questions that require about 23 minutes each to answer and are worth 10 points each. AP Chemistry 2019 Free-Response Questions Le Chatelier Lab-2/14/13 Purpose To observe how the system responds to various stresses, such as changes in the concentration of reactants or products, or in the temperature of the system.

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Circle your answer below. Justify your answer and include a calculation of  $E^\circ$  for the overall reaction.  $\text{H}_2\text{O}(\text{aq}) + \text{Na}_2\text{S}(\text{aq}) + \text{Na}_2\text{SO}_3(\text{aq}) \rightarrow \text{Na}_2\text{S}_2\text{O}_3(\text{aq}) + \text{H}_2\text{O}(\text{aq})$  [Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>(aq) should be circled.] The reaction between S<sub>2</sub>O<sub>3</sub><sup>2-</sup>(aq) and I<sub>2</sub>(s) will be thermodynamically favorable because of the reaction is positive ( $\Delta G^\circ = 0.54 - 0.08 = +0.46 \text{ V}$ ), from which it

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